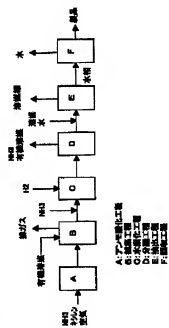


<p>2003-407667/39 A41 E14          MITSUBISHI GAS CHEM CO INC          2001.07.16 2001-215005(A+2001IP-215005) (2003.01.29) C07C          209/48, 21/27 // C07B 61/00  <b>Manufacture of high purity xylene diamine used as raw material for polyamide resin, involves adding water and specific solvent, to rough xylene diamine, and recovering high purity xylene diamine</b>          C2003-108817</p>	<p>MITN 2001.07.16          *JP 2003026638-A          A(1-E5, 8-D3) E(10-B1E, 11-Q1) N(2-A, 2-B, 2-C, 3-C1, 3-D2)          contacted with an organic solvent, directly to collect the phthalonitrile in the organic solvent. Subsequently, liquid ammonia is added to the organic solvent, without separating the phthalonitrile, and hydrogenation reaction is performed. Subsequently, organic solvent and ammonia are separated from the hydrogenation reaction product, to obtain rough xylene diamine. Water and at least one type of solvent chosen from aromatic hydrocarbon and saturated hydrocarbon, are added to rough xylene diamine, to separate solvent phase and water phase. Subsequently, high purity xylene diamine is recovered from the water phase to which extraction-separation is performed. Phthalonitrile is synthesized by ammoxidation reaction of metaxylene or paraxylene.</p>
<p><b>NOVELTY</b>          Xylene is ammoxidated to obtain phthalonitrile. The ammoxidated gas is contacted with organic solvent. Liquid ammonia is added to organic solvent, and hydrogenation is performed. Organic solvent and ammonia are separated from hydrogenation reaction product. Water and aromatic hydrocarbon and/or saturated hydrocarbon solvent, are added to resulting rough xylene diamine (XD), and high purity XD is recovered.</p>	<p><b>USE</b>          For manufacture of high purity xylene diamine used as raw material for polyamide resin and epoxy hardener, and as intermediate raw material of isocyanate.</p>
<p><b>DETAILED DESCRIPTION</b>          Ammoxidation of raw material xylene is performed by gaseous-phase contact reaction with ammonia and oxygen-containing gas, to obtain a phthalonitrile. The resulting ammoxidation reactive gas is</p>	<p><b>ADVANTAGE</b>          JP 2003/026638-A+</p>

<p>Xylylene diamine of high purity is obtained efficiently with sufficient yield.</p> <p><b>DESCRIPTION OF DRAWING</b></p> <p>The figure shows the flowchart of manufacturing method of high purity xylylene diamine. (Drawing includes non-English language text).</p> <p><b>EXAMPLE</b></p> <p>Ammonoxidation of metaxylylene was performed by gaseous-phase contact reaction with ammonia and oxygen-containing gas, and phthalonitrile was obtained. The resulting ammonoxidation reactive gas was contacted with pseudo cumene organic solvent, and the phthalonitrile was collected in the organic solvent. Subsequently, liquid ammonia was added to the organic solvent, without separating the phthalonitrile, and hydrogenation reaction was performed. Subsequently, organic solvent and ammonia were separated from the hydrogenation reaction product, and rough metaxylylene diamine was collected. Water was added to the rough metaxylylene diamine, to separate solvent phase and water phase. Subsequently, high purity metaxylylene diamine was recovered from the water phase. Batch distillation of metaxylylene diamine containing water phase was</p>	<p>performed, and water was separated, and high purity xylylene diamine with purity of 99.99 wt.%, was obtained.</p> <p><b>TECHNOLOGY FOCUS</b></p> <p>Organic Chemistry - Preferred Solvent: The organic solvent is an aromatic hydrocarbon. Preferred Catalyst: Manufacture of high purity xylylene diamine is performed using a fluid catalyst containing one or more types of metallic oxide chosen from vanadium, molybdenum and iron. Hydrogenation is performed in presence of nickel and/or cobalt catalyst.</p>
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